

# **2. Atoms, molecules and stoichiometry**

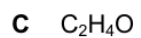
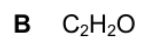
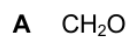
## **2.3 Formulas**

### **Paper 1**

Question Paper

**1** Compound X consists of 40.0% carbon, 6.7% hydrogen and 53.3% oxygen by mass.

What is the empirical formula of compound X?



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